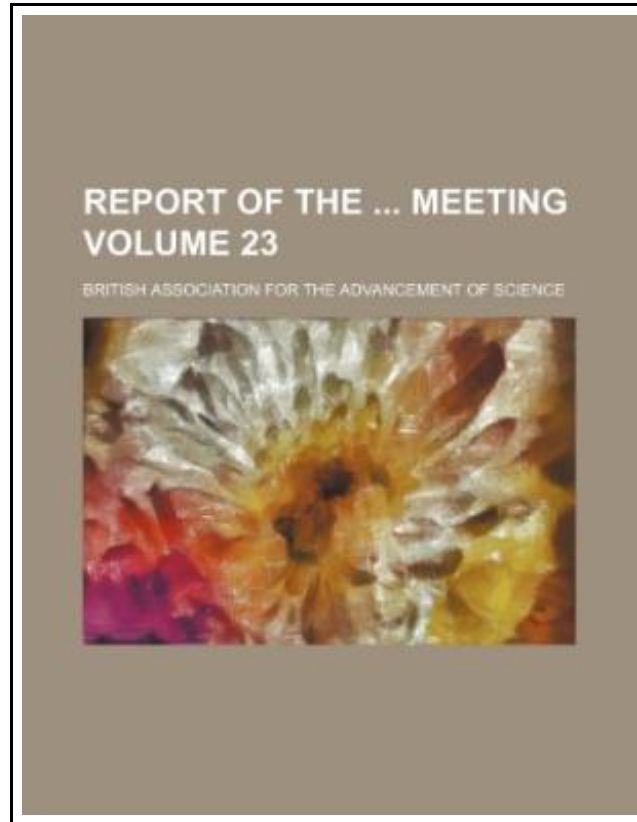


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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1854 Excerpt: .of Water under Pressure, by Ilie Galvanic Battery. By John P. Gassiot, V.P.R.S. It is a well-known law, long since discovered by Dr. Faraday, that whatever may be the liquid electrolysed by the galvanic battery, and whatever may be the size of the electrodes used, the same amount of chemical decomposition takes place in each cell of the battery; if water is placed in any number of separate vessels, connected by platinum electrodes with each other, and thus introduced within the circuit of a galvanic battery, the same amount of the mixed gases will be evolved from each vessel. If, instead of allowing the gases to escape from each, such vessels are closed, a very considerable amount of pressure is obtained. The late Professor Daniel many years since experimented with a closed glass vessel, as did Dr. Leeson, but in each instance, before any great pressure was obtained, the vessels broke with very considerable violence. Will Faraday s law hold good under extreme pressure? Is there any point at which the pressure will recombine the hydrogen and oxygen evolved by the electrolysis of water? Is there any point at which water under such pressure will cease to be electrolysed? and will it, under such circumstances, continue a conductor? All these questions appear to me to be well worthy of examination; and although the experiments I have hitherto made are far from being conclusive, they prove that, as far as I have been able to obtain apparatus...



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